Ca	ambridge Lo	ower Secondary Che	eckpoint
CANDIDATE NAME			
SCIENCE			1113/01
Paper 1			April 2020
			45 minutes
Candidates answ	er on the Question Pa	per.	
Additional Materia	als: Pen Pencil Ruler	Calculator	

1 The list contains the names of different parts of a cell.



cell membrane cell wall chloroplast

Complete the table by placing ticks (\checkmark) in the correct boxes.

	where the part of the cell is found			
part of a cell	only in animal cells	only in plant cells	in both animal and plant cells	
cell membrane				
cell wall				
chloroplast				

[2]

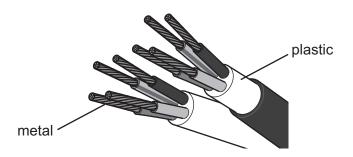
2 This question is about the properties of metals.



(a) Draw straight lines to match the **property of a metal** with its correct **use** linked to that property.

[2]

(b) Metals are used to make electrical wires because they conduct electricity.

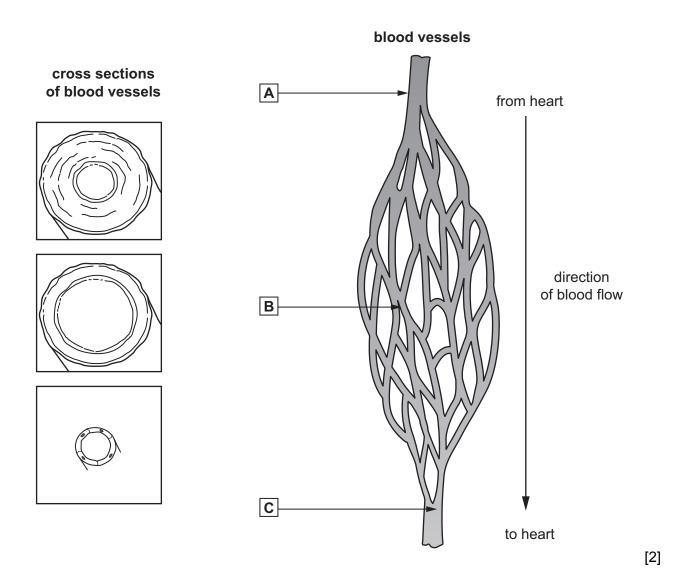


Write down **two** reasons why plastic is put around electrical wires.

1	
2	

	(c) Read the sentences about the physical properties of metals.	
	Tick (\checkmark) the box next to the correct sentence.	
	All metals have low melting points.	
	Some metals are gases at room temperature.	
	All metals conduct heat.	
	All metals are brittle.	[4]
3	Blessy uses the internet to find out about our Solar System.	[1]
W	She finds a very old model of our Solar System.	
	Jupiter Mars orbits Earth Moon Venus Saturn	
	Scientists thought that the Earth was at the centre.	
	The orbits in the old model show the other objects moving around the Earth.	
	(a) Scientists today know that the Earth is not at the centre of our Solar System.	
	What is at the centre of our Solar System?	
		[1]
	(b) Write down two other things that are incorrect in the old model.	
	1	
	2	[2]
	(c) Write down one thing that is correct in the old model.	
		[1]

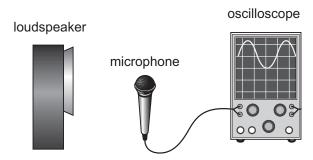
- 4 This question is about blood vessels.
- R
- (a) Draw a straight line from each **cross section of a blood vessel** to the correct **letter** showing where the blood vessel is found.



(b)	Name the types of blood vessel labelled A and C .

Α	
С	

The	e picture shows three different eleme	ents and their state at room to	emperature.
	solid iodine	liquid bromine	chlorine gas
(a)	Which two of these elements flow	easily at room temperature?	
			[1]
/L-\	Will at its the sale and a law and all for all	La via a O	
(D)	What is the chemical symbol for ch	liorine?	
			[1]
(c)	Chlorine gas fills the jar.		
	Explain why a gas fills a jar.		
	Tick (✓) the box next to the correct	explanation.	
	Forces between the particles p	ush them apart.	
	The particles are free to move.		
	The particles can easily be squ	ashed into a small space.	
	The particles increase in size to	o fill the space.	
(d) l	iquid bromine easily evaporates.		[1]
E	Explain what happens to the particles	s (molecules) when a liquid e	vaporates.
_			
•			······
•			

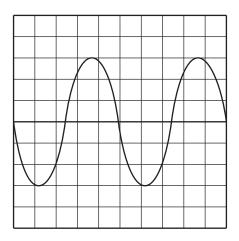


She makes a sound using a loudspeaker.

The sound is detected by the microphone.

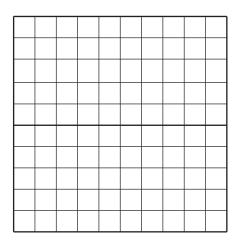
(a)	Describe now the sound travels from the loudspeaker to the microphone.
(a)	Describe how the soling travels from the lougsheaker to the microphone

(b) Mia draws the wave she sees on the oscilloscope.



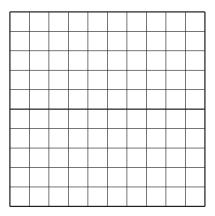
(i) She makes the sound louder.

Draw this wave on the oscilloscope.



(ii) She makes the sound a higher pitch.

Draw this wave on the oscilloscope.



7 This question is about the life cycle of a plant.

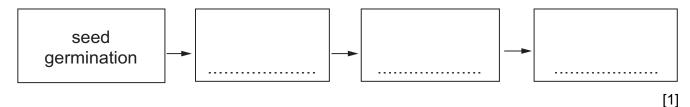


(a) These processes take place in the life cycle of a plant.

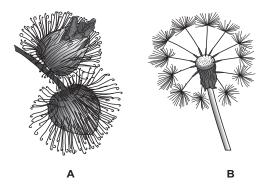
fertilisation pollination seed formation seed germination

Put the processes in the order in which they occur in the life cycle of a plant.

One has been done for you.



(b) The diagrams **A** and **B** show two different types of seed.

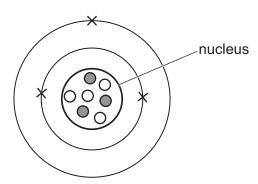


Suggest the method of dispersal for each type of seed.

Give a reason for each answer.

	method of seed dispersal	reason
A		
В		

[1]



- (a) There are three electrons in a lithium atom.
 - (i) How many protons are there in a lithium atom?

[1]	1
 ٠	,

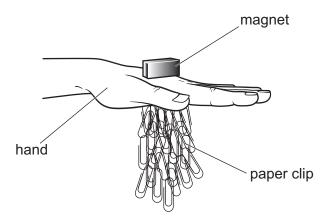
(ii) How many neutrons are there in a lithium atom?

F41
- 111
F.1

(b) A sodium atom contains 11 protons.

Draw the structure of a sodium atom.





Mike

- puts the magnet on top of his hand
- puts the bottom of his hand onto 24 paper clips
- lifts his hand up
- counts how many paper clips have been attracted
- repeats with different magnets.

Here are his results.

magnet	number of paper clips
Α	24
В	24
С	7
D	19
E	12

(a)	Mik	e concludes that	magnet A and	magnet B are both strong.	
	He cannot conclude which of these two magnets is stronger.				
	(i)	Explain why he	cannot conclu	de which magnet, A or B , is	stronger.
					[1]
	(ii)	Describe what he	e could do to f	ind out which magnet, A or I	3 , is stronger.
					[1
(b)	Pie	rre repeats Mike's	s investigation.		
	Her	re are Pierre's res	ults.		
			magnet	number of paper clips	
			A	18	
			В	22	-
			С	1	
			D	13	
			E	6	
	Pie	Pierre uses the same magnets as Mike.			
	Pierre uses the same paper clips as Mike. The results are different.				
	(i)	Describe one dif	ference betwe	een the results.	
				[1]	
(ii) Suggest why the results are different.					
					[1]

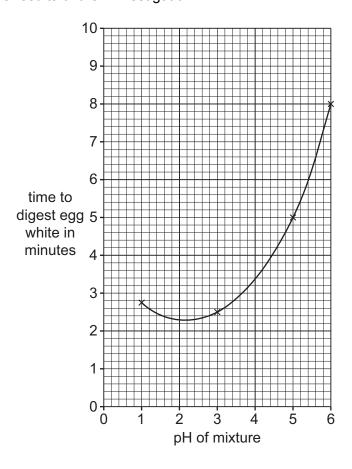
10 Egg white contains a protein.



An enzyme digests protein in the stomach.

Class 9 investigate how changing the pH affects the time it takes for the enzyme to digest egg white.

The graph shows the results of their investigation.



(a)	(i)	Which pH has the shortest time of digestion?	
		pH	[1]
	(ii)	The class want to be certain that they have found the shortest time.	
		Describe two things the class does to be certain.	
		1	
		2	

(b)	(i)	State one safety risk of using liquids with a very low pH.	
			[1]
	(ii)	Describe one way of reducing this safety risk.	
			[1]
(c)	Sta	te one variable the students must control in this investigation.	

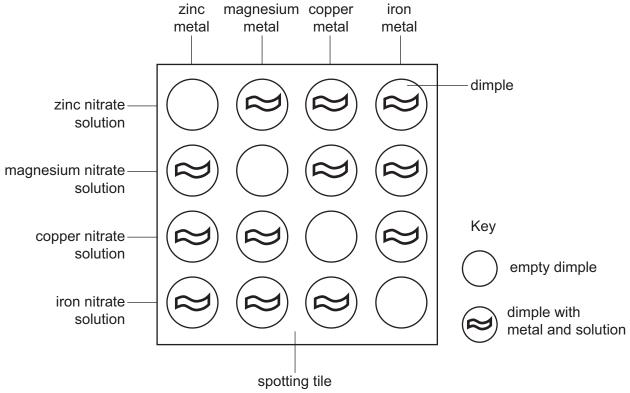
11 Jamila and Ahmed investigate displacement reactions.



They put drops of different solutions into the dimples of a spotting tile.

They then add metals to each solution.

The diagram shows their experiment.



	spotting tile	
(a)	Jamila and Ahmed look to see if a reaction takes place.	
	Suggest what they might see if a reaction takes place.	
		[1]

(b) They record their results in a table.

They put a

- tick (✓) if there is a reaction
- cross (x) if there is no reaction.

Here are some of their results.

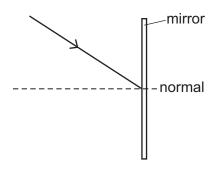
adution	metal				
solution	zinc	magnesium	copper	iron	
zinc nitrate		✓	Х	Х	
magnesium nitrate					
copper nitrate	✓	✓		✓	
iron nitrate	✓	✓	Х		

(i)	Complete the table to predict the results for magnesium nitrate.	[1]
(ii)	The reactivity series shows the metals in order of reactivity.	
	Which of the four metals is the lowest in the reactivity series?	
		[1]

12 Complete the light rays in the three diagrams.

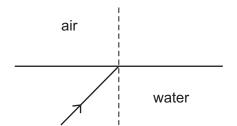


(a) reflection



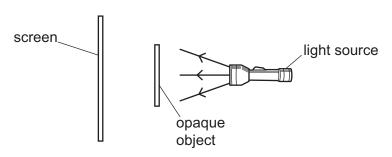
[1]

(b) refraction



[1]

(c) shadow formation



[2]

13 Complete the word equation for aerobic respiration.



