

Cambridge Lower Secondary Checkpoint

SCIENCE		1113/01
CENTRE NUMBER	CANDIDATE NUMBER	
CANDIDATE NAME		



You must answer on the question paper.

No additional materials are needed.

INSTRUCTIONS

Paper 1

- Answer all questions.
- Use a black or dark blue pen. You may use an HB pencil for any diagrams or graphs.
- Write your name, centre number and candidate number in the boxes at the top of the page.
- Write your answer to each question in the space provided.
- Do not use an erasable pen or correction fluid.
- Do not write on any bar codes.
- You should show all your working in the booklet.
- You may use a calculator.

INFORMATION

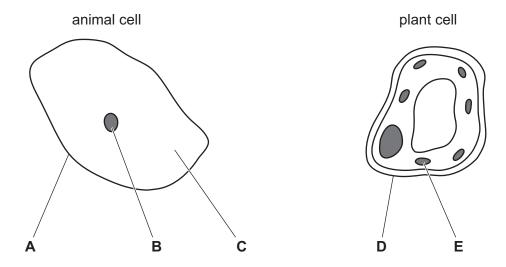
- The total mark for this paper is 50.
- The number of marks for each question or part question is shown in brackets [].

This document has **16** pages. Any blank pages are indicated.

October 2022 45 minutes 1 The diagram shows an animal cell and a plant cell.



Parts of the cells are labelled with the letters A, B, C, D and E.



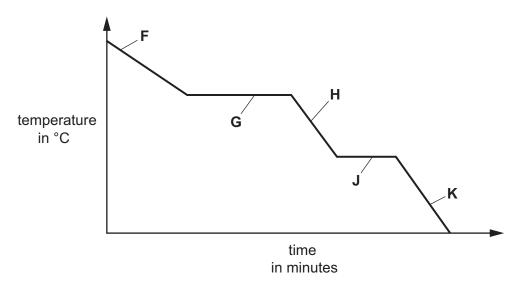
Draw a straight line from each **letter** to the **name** of the part of the cell.

letter	name
Α	cell membrane
В	cell wall
С	chloroplast
D	cytoplasm
Е	nucleus

[4]

2 The diagram shows a cooling curve for a pure substance.





(a) The table contains descriptions for each letter in the diagram.

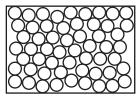
Write the correct letter next to each description.

One has been done for you.

description	letter
gas	F
solid	
condensing	
freezing	
liquid	

[2]

(b) The diagram shows the particles at point **H** on the cooling curve.



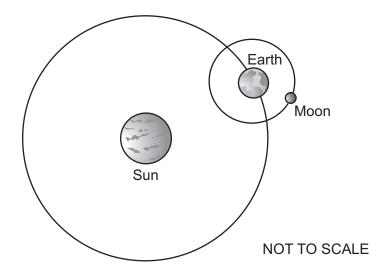
Draw the particles at point **F** in this box.



[1]

3 Look at the diagram of the Sun, the Earth and the Moon.





(a) We can see both the Sun and the Moon from the Earth.

Complete the sentences.

We can see the Sun because the Sun	light.
We can see the Moon because light from the	
isto the	Earth.

(b) We have both day and night on the Earth.

Explain how both day and night happen.

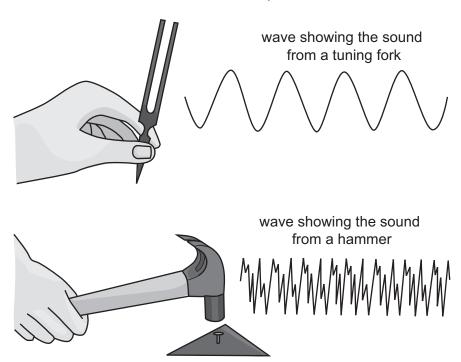
[2]

[2]

4	(a)	Lool	k at the fiv	ve processes th	at take place during p	lant reproduction.		
R		The	y are not	in the correct o	rder.			
		disp	ersal	fertilisation	seed formation	pollen formation	pollination	
		Com	plete the	table to show t	he correct order of the	ese five processes.		
		One	has beer	n done for you.				
					fertilisation			
								[2]
	(b)	Sug	gest with	a reason how t	he seeds of an orange	e tree are dispersed.		
		sugg	gestion					
		reas	on					
								[3]
5	Sul	fur is	an eleme	ent				
®	(a)				mbol for sulfur.			[1]
	(a)	VVII	ie down i	ne chemical syl	indoi ioi suiiui.		•••••	ניו
	(b)	Sulfi	des are c	compounds that	contain sulfur.			
		(i)	Write dov	wn the element	that all chlorides cont	ain.		[1]
		(ii)	A compo	und has this ch	emical formula.			
					NaOH			
			Write dov	wn the names o	of the three elements i	in this compound.		
			1					
			•					
			·					[2]

6 Chen compares sound waves seen on an oscilloscope.



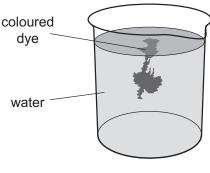


The loudness of sound from the tuning fork and the hammer are the same.				
Describe how the waves show that the sounds are the same loudness .				
[1]				
Describe two differences between the sound waves.				
1				
2				

[2]

The diagram shows what happens to a coloured dye after 10 minutes.









after 10 minutes

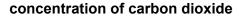
Explain the changes seen in the diagram.

Use ideas about particles.

		ro.
		L ²

Hydrogencarbonate indicator is a harmless coloured liquid used to estimate the concentration of carbon dioxide in water.

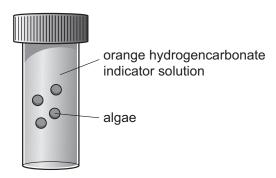
colour of hydrogencarbonate indicator





The diagram shows some algae in a bottle of orange hydrogencarbonate indicator.

Algae are small green plants.



(a) Why do green plants need carbon dioxide?

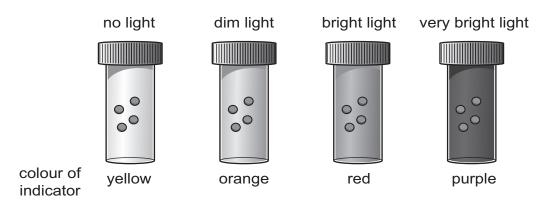
F.47	
111	
լ լ լ	

(b) Aiko investigates the effect of light intensity on the algae.

She puts bottles of algae and orange hydrogencarbonate indicator in different light intensities.

She leaves the bottles in the light for two hours.

The diagram shows the colours of the hydrogencarbonate indicator after two hours.



As the light intensity increases the concentration of carbon dioxide in the bottles changes.

Describe how the concentration of carbon dioxide changes.

Explain your answer.

	•	•		
	des	cription		
	exp	lanation		
				[2]
				L
(c)	(i)	Why is i	it important to use the same amount of algae in each jar?	
				[1]
	(ii)	State tw	vo other variables Aiko keeps the same in this investigation.	
		1		
		2		
				[2]
(d)		en a bot inges col	tle containing algae is kept in the dark the orange hydrogencarbonate indicaour.	ator
	Sug	gest why	y.	
				[1]

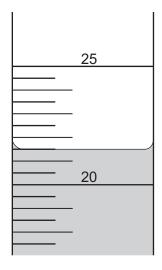
9 Carlos investigates the temperature change during some reactions.



Carlos:

- measures 20 cm³ of a liquid in a measuring cylinder
- puts the liquid in a plastic cup and measures the temperature of the liquid
- adds a solid to the liquid and stirs
- measures the temperature of the mixture
- repeats this experiment with different solids and different liquids.
- (a) Carlos puts too much liquid in the measuring cylinder.

Look at the diagram of the measuring cylinder and the liquid.



How much liquid does Carlos remove to get the $20\,\mathrm{cm}^3$ he needs?

	cm^3	[1]
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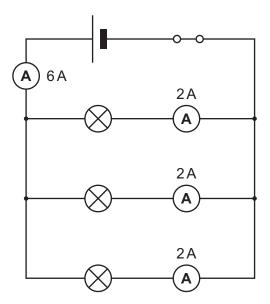
(b) Here are his results.

liquid used	temperature of liquid at start in °C	solid added	temperature of mixture at end in °C	change in temperature in °C	is the reaction exothermic or endothermic?
copper sulfate solution	19	magnesium powder	30		
dilute ethanoic acid	19	sodium carbonate	12	-7	
potassium carbonate solution	18	citric acid	14		
dilute sulfuric acid	18	magnesium ribbon	34		

(i) Calculate the change in temperature for each reaction.				
		One has been done for you.		
		Write your answers in the table.	[1]	
	(ii)	Which mixture of liquid and solid releases the most energy?		
		liquid		
		solid		
			[1]	
	(iii)	Complete the table by writing endothermic or exothermic in the last column.	[1]	
(c)	Cai	Carlos is not sure all his results are reliable.		
	Wh	at does Carlos do to make his results more reliable?		
			[1]	

10 Jamila connects an electrical circuit.





She uses **four** ammeters in the electrical circuit.

There is an ammeter reading next to each ammeter symbol.

(i)	What type of electrical circuit is shown in the diagram?	
		[1]
(ii)	Explain what is happening in the electrical circuit.	
	Use the readings on the ammeters in your answer.	
	· · · · · · · · · · · · · · · · · · ·	
		••••
		[2]
Writ	te down the name of the equipment that measures voltage .	
		[1]
(ii)	ii) Explain what is happening in the electrical circuit. Use the readings on the ammeters in your answer.

11 Mia feels unwell and visits the doctor.

R	She	complains that she ha	s a lack of energy.					
	The doctor says this is because there is not enough haemoglobin in her blood.							
This is caused by her diet. (a) What term describes the type of disease caused by diet?								
	deficiency disease developmental disease							
genetic disease								
			infectio	us disease		[1]		
	(b)	Explain why not havin	g enough haemog	obin in the blood cause	s a lack of energy.			
						••••		
						[2]		
(c) Circle the element needed to make haemoglobin.								
		calcium	iron	potassium	sodium	[1]		

14 **12** Hassan finds this timeline of life on Earth on the internet. The timeline is not drawn to scale. present day ----65 million years ago — extinction of the dinosaurs

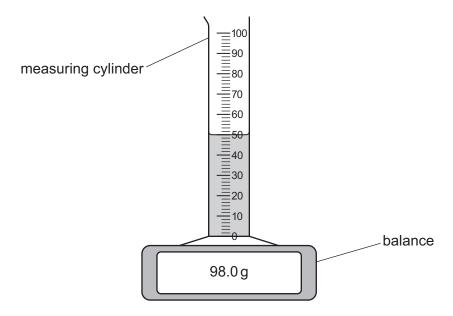
420 million years ago ———	first fish				
450 million years ago ———	first land plants				
3500 million years ago ———	first life on Earth				
4500 million years ago ———	Earth formed				
more than 10 000 million years ago ———	Universe formed				
(a) What is the age of the Earth?	million years [[1]			
(b) How long after the Earth was formed did the first land plants appear?					
	million years	[1]			

13 Blessy calculates the density of a liquid.



She puts 50 cm³ of the liquid in a measuring cylinder.

She puts the measuring cylinder on a balance.



(a) The unit of density is g/cm³.

Write down the equation to calculate density.

density =	[1	[]
	L	1

(b) The mass of the empty measuring cylinder is 38.0 g.

The mass of the measuring cylinder and the liquid is 98.0 g.

(i) Calculate the mass of the liquid.

mass of liquid [1] _____g

(ii) The volume of the liquid is 50 cm³.

Calculate the **density** of the liquid.

density of the liquid [1]