

#### **Cambridge Lower Secondary Checkpoint**

SCIENCE 0893/02

Paper 2 April 2023

MARK SCHEME

Maximum Mark: 50

#### **Published**

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Markers were instructed to award marks. It does not indicate the details of the discussions that took place at a Markers' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the End of Series Report. Cambridge will not enter into discussions about these mark schemes.

Question	Answer	Marks	Further Information
1(a)	DNA <b>or</b> deoxyribonucleic acid	1	Do not accept chromosome
			Ignore genes or genetic material or allele
1(b)(i)	2 white <b>and</b> 3 grey	1	
	or		
	1 white <b>and</b> 4 grey		
	or		
	0 white <b>and</b> 5 grey		

Question	Answer	Marks	Further Information
1(b)(ii)	any three from	3	each correct answer = 1 mark
			Accept ora where appropriate
			Accept black or brown for grey throughout
	M1 (idea that) less predation of grey shrews or fewer grey shrews eaten or grey shrews are more difficult to hunt		Accept idea that fewer grey shrews are killed
	<b>M2</b> grey shrews have better camouflage <b>or</b> (idea that) grey ones blend in with the ground better than white shrews		Accept (idea that) easier to see the white shrews
	<b>M3</b> (idea that) more grey shrews survive <b>or</b> so grey shrews number increases		
	<b>M4</b> more grey shrews are able to breed <b>or</b> more grey shrews reproduce		
	<b>M5</b> pass grey (fur) gene onto next generation <b>or</b> pass advantageous gene onto next generation		<b>Accept</b> pass on advantageous mutation to next generation
	M6 (process) repeats for many generations		

Question	Answer	Marks	Further Information
1(b)(iii)	any two from	2	each correct answer = 1 mark
	wear gloves (handling shrews)		Accept use nets (instead of hands)
	eye protection		Accept glasses or goggles
	(face) mask <b>or</b> filter		Ignore protective suit
	disinfect traps (between use) <b>or</b> sanitise (traps)		Accept wash hands after investigation or touching shrews

Question	Answer	Marks	Further Information
2(a)	2.8.7	1	Accept a diagram of the correct electronic structure
2(b)	Mg	1	Accept magnesium, but symbol takes precedence
2 (c)	any one from	1	
	unreactive <b>or</b> inert		Do not accept little reaction
	gas at room temperature		Accept low density
			Note being a gas is not sufficient
	low melting point <b>or</b> low boiling point		Accept has a higher boiling point than neon or has a higher melting point than neon
			Accept atom has 18 protons or proton number is 18 or atomic number is 18 or has a stable electronic structure or has a stable atomic structure or it has eight electrons in its outer shell
	colourless		Do not accept is coloured or colourful
			Do not accept is a metal or a halogen

Question			Answer	Marks	Further Information
3(a)	_			1	all four correct for the mark
		part of plant	order of pathway		
		leaf	(5)		
		leaf xylem	4		
		root hair cell	1		
		root xylem	2		
		stem xylem	3		
3(b)	transpi	iration		1	Accept evaporation or evapotranspiration  Ignore diffusion
					Do not accept sweating
3(c)	reduce	es population size <b>or</b> p	opulation declines	1	Ignore they die or cannot grow as well or becomes extinct
					<b>Ignore</b> any reason given

Question	Answer	Marks	Further Information
4(a)	w	1	more than <b>one</b> answer circled = 0 marks <b>Accept</b> any indication of the correct answer, e.g.ticking or underlining, but circling takes precedence
4(b)	(number of waves) 4	1	
4(c)	(time = 1 second) reinforce  (time = 3 seconds) cancel each other <b>or</b> cancel out	2	each correct answer = 1 mark  Accept constructive interference  Accept (produces a) louder sound or amplify the sound or larger amplitude  Accept destructive interference  Accept (produces a) quieter sound or smaller amplitude

Question	Answer	Marks	Further Information
5(a)	respiration	1	Do not accept breathing
5(b)	photosynthesis	1	
5(c)	both release carbon dioxide <b>or</b> both release heat (into the surroundings) <b>or</b> both exothermic	1	Ignore they are both part of the carbon cycle
			Ignore they break down carbon (compounds)
			Do not accept both release carbon

Question	Answer	Marks	Further Information
5(d)	any two from	2	each correct answer = 1 mark
			<b>Ignore</b> reference to ozone and ozone depletion
	ice caps melt <b>or</b> glaciers melt		Note ice melts or snow melts unqualified is not sufficient
	sea levels rise		Accept sea level changes
	(more) flooding <b>or</b> (more) land under water <b>or</b> (more) cities under water		
	drought <b>or</b> famine		
	(more frequent) extreme weather, e.g. (more) typhoons/cyclones/tornadoes		Accept (more extreme) heat waves
			Ignore weather change or unpredictable weather
	global warming		Accept Earth or atmosphere becomes warmer
			Accept sea temperatures rise or ocean warming
			Accept temperature change or temperature increase
	extinction of plant (species) <b>or</b> extinction of animal (species)		Accept examples of extinction, e.g. danger of polar bears becoming extinct or mass extinction (of plants and/or animals) or coral bleaching or loss of habitats
	(increased number of) forest fires <b>or</b> wildfires		

Question	Answer	Marks	Further Information
6(a)	(as you go down the group atomic radius) increases <b>or</b> goes up	1	
6(b)	number in the range 20 to 45 (°C)	1	Accept a range of values providing the range is entirely within 20 to 45 °C  Accept answer written in table, but answer line takes precedence
6(c)	giant (ionic)	1	Do not accept giant covalent  Ignore lattice

Question	Answer	Marks	Further Information
7		2	each correct answer = 1 mark
	buzzer		
	variable resistor		Accept rheostat
			Do not accept resistor on its own
			Do not accept variation resistor

Question	Answer	Marks	Further Information
8(a)	C B (E) D (F) A	3	all four letters in the correct order = 3 marks
			CB in correct order = 1 mark
			<b>D</b> in the middle = 1 mark
			A at the end = 1 mark
8(b)	rocks from the Earth and the Moon had the same properties <b>or</b> Moon rocks have a lower density than Earth rocks	1	Accept rocks from the Moon were similar to rocks on Earth or rocks from the Moon are same type as rocks on Earth or has the same components/composition as rocks on Earth or Moon and Earth contain the same minerals  Do not accept Moon and Earth have an iron core
8(c)	any one from	1	Accept ora
	(idea that) the dense iron from the cores (of both planets) merged to create the Earth		
	there are more dense <b>rocks</b> on the Earth		
	the less dense rocks (were ejected and) formed the Moon		<b>Accept</b> it was the less dense rocks that were ejected

Question	Answer	Marks	Further Information
9(a)		2	each correct tick = 1 mark
	changes in seasons		three ticks and two correct = 1 mark
			three ticks and one correct = 0 marks
	changes to the environment over time		four or five ticks = 0 marks
	increased reproduction		Accept any indication of the correct answer, e.g. circling or underlining, but ticking takes precedence
	new diseases		
	new food sources		
9(b)	(idea that) death rate greater than reproduction rate	1	Accept low reproduction rate and high death rate
			Accept death rate is increasing and the reproduction rate is decreasing
			Accept birth rate for reproduction rate
			Ignore reference to no reproduction

Question	Answer	Marks	Further Information
10(a)	displacement	1	more than <b>one</b> answer circled = 0 marks <b>Accept</b> any indication of the correct answer,
			e.g. ticking or underlining, but circling takes precedence
10(b)	copper + silver nitrate → copper nitrate + silver	1	Accept reactants in either order and products in either order
			Accept = for →
			<b>Accept</b> correct symbol equation (balanced or unbalanced), but names take precedence
			Cu + $2AgNO_3 \rightarrow Cu(NO_3)_2 + 2Ag$

Question	Answer	Marks	Further Information
10(c)		2	each correct answer = 1 mark
	copper + iron nitrate		more than <b>one</b> answer ticked = 0 marks for the question
	magnesium + zinc nitrate ✓		incorrect answer ticked = 0 marks for the question
	silver + magnesium nitrate		Accept any indication of the correct answer, e.g. circling or underlining, but ticking takes precedence
	iron + zinc nitrate		
	magnesium is more reactive than zinc  or  magnesium is higher in the reactivity series than zinc  or  zinc is less reactive than magnesium  or  zinc is lower in the reactivity series than magnesium		Note magnesium is more reactive is not sufficient or it is more reactive than zinc is not sufficient  Note the more reactive metal displaces the less reactive metal is not sufficient must refer to magnesium and zinc  Do not accept magnesium is more reactive than zinc nitrate

Question	Answer	Marks	Further Information
11(a)	°C	1	Accept degrees C or degrees Celsius or °Celsius or Kelvin(s) or K
			<b>Accept</b> answer written in table(s), but answer line takes precedence
			Do not accept C°
11(b)	(cardboard) 18 (°C)	1	all three correct for the mark
	(newspaper) 17 (°C)		
	(bubble wrap) 12 (°C)		
11(c)	yes (no marks)	1	<b>Note</b> if no = 0 marks for the question
	the temperature has decreased by the smallest amount <b>or</b> the temperature drop is the smallest (over the same amount of time) <b>or</b> least temperature change		<b>Accept</b> temperature change was only 12 °C compared to 17 °Cand 18 °C
11(d)		2	<b>Note</b> if any other temperature <b>or</b> time selected = 0 marks for the question
			each correct answer = 1 mark
	67 circled in the table for cardboard		Accept 67 written by the question, but circling in the table takes precedence
			Accept 60 circled in the table for cardboard
	(idea that) 67 does <b>not</b> fit the pattern <b>or</b> 67 not in the correct sequence		<b>Accept</b> temperature drops then gets higher <b>or</b> (idea that) temperature at 60 s is less than at 120 s <b>or</b> temperature should be higher than 75 °C

Question	Answer	Marks	Further Information
11(e)	any two from	2	each correct answer = 1 mark
	use the same starting temperature		
	use the same temperature of the surroundings <b>or</b> same room temperature		
	use the same volume of water		<b>Accept</b> use same mass of water <b>or</b> same amount of water <b>or</b> same level of water
	use a lid on each beaker		<b>Accept</b> cover the top of beaker <b>or</b> put a stopper on the beaker
			<b>Ignore</b> put a layer of oil on water so heat doesn't escape
	use the same thickness of insulation		Accept same surface area of insulation
	repeat the investigation		Accept repeat the investigation and calculate the mean = 2 marks
	stir the water		Accept greater variety of insulation
			Accept insulate the bottom of the beaker
			Accept smaller time intervals between readings
			<b>Accept</b> use a temperature sensor and a data logger
			<b>Ignore</b> a more accurate or precise thermometer

Question	Answer	Marks	Further Information
12(a)	(gas) syringe <b>or</b> measuring cylinder (inverted in a trough of water)	1	Accept graduated cylinder
12(b)	repeats (the investigation)	1	<b>Ignore</b> changing time intervals <b>or</b> using a different total time
12(c)	(safety risk) hydrochloric acid is corrosive	2	each correct answer = 1 mark  Note award the answers wherever they are seen  Accept acid or hydrochloric or solution (in the flask) for hydrochloric acid  Accept hydrochloric acid burns or hydrochloric acid is harmful or hydrochloric acid getting into eyes or hydrochloric acid is an irritant  Note hydrochloric acid coming into contact with hands is not sufficient  Ignore references to the gas being produced or explosions
	(how the risk is reduced) wear goggles <b>or</b> gloves		Do not accept hydrochloric acid is flammable  Accept gloves or apron or lab coat or glasses
			Ignore facemask or protective gear